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# Magnetic structure of the YbMn<sub>2</sub>SbBi compound

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# ABSTRACT

A neutron diffraction investigation has been carried out on the trigonal La<sub>2</sub>O<sub>2</sub>S-type (hP5, space group  $P\bar{3}ml$ , No. 164; also CaAl<sub>2</sub>Si<sub>2</sub>-type) YbMn<sub>2</sub>SbBi intermetallic compound. The YbMn<sub>2</sub>SbBi presents antiferromagnetic ordering below 138(3)K and ferrimagnetic ordering below 112(3)K. Between 138 and 112 K, the magnetic structure of YbMn<sub>2</sub>SbBi consists of antiferromagnetically coupled *ab*-plane magnetic moments of the manganese atoms ( $\mathbf{D}_{1d}$  magnetic point group). Below 112(3)K, the ferromagnetic components of Yb and Mn begin to develop, and the magnetic structure of YbMn<sub>2</sub>SbBi becames the sum antiferromagnetic component with  $\mathbf{D}_{1d}$  magnetic point group and ferromagnetic one with  $\mathbf{C}_2$  magnetic point group. The magnitude of Yb and Mn magnetic moments in YbMn<sub>2</sub>SbBi at 2 K ( $M_{Yb}$  = 3.6(2)  $\mu_B$ ,  $M_{Mn}$  = 3.5(2)  $\mu_B$ ) correspond to the trivalent state of the Yb ions and tetravalent state of the Mn ions.

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# 1. Introduction

The existence of the YbMn<sub>2</sub>Sb<sub>2</sub> compound, adopting the La<sub>2</sub>O<sub>2</sub>Stype structure (also called CaAl<sub>2</sub>Si<sub>2</sub>-type; hP5, Space group  $P\bar{3}ml$ , No. 164,  $\mathbf{D}_{3d}$  ( $\overline{3}2/m$ ) point group) was earlier reported by Ruhl and Jeitschko [1]: in this structure prototype the Yb atoms occupy the 1a(0, 0, 0) special position, manganese atoms occupy the 2d site  $Z_{\rm Sb}$ ). The magnetic ordering (with a magnetic transition at about 120 K) and magnetic structure have been found for this compound in works [2,3]. The commensurate magnetic structure of YbMn<sub>2</sub>Sb<sub>2</sub> (C<sub>i</sub> magnetic point group, P1 magnetic space group) consists of antiferromagnetically coupled manganese magnetic moments in the Mn1 [1/3, 2/3, Z<sub>Mn</sub>] and Mn2 [2/3, 1/3, -Z<sub>Mn</sub>] positions, the ytterbium magnetic moment is zero. At 4 K, the magnetic moment of the Mn atom is  $\mu_{Mn} = 3.6(1) \mu_B$  [3] that corresponds to the tetravalet state of manganese in YbMn<sub>2</sub>Sb<sub>2</sub> ( $M^{Mn4+} = 4 \mu_B [4]$ ).

The substitution of Sb for Bi in YbMn<sub>2</sub>Sb<sub>2</sub> compound leads to the formation of the YbMn<sub>2</sub>SbBi solid solution with changes in interatomic distances and considerable changes in the magnetic parameters. Hence, to determine the type of magnetic ordering and the magnetic structure of YbMn<sub>2</sub>SbBi, neutron diffraction study was carried out and the results are presented here.

#### 2. Synthesis and experimental details

Commercial ytterbium (pieces cut from ingot, with purity 99.9 wt.%), manganese (small grains from a platelet previously surface-cleaned by conc. HNO<sub>3</sub>, with purity 99.99 wt.%), antimony (grains, with 99.999 wt.% purity) and bismuth (grains, with 99.999 wt.% purity), were used as the starting components. After few preparations were firstly attempted in an arc-furnace, and due to the fact that preparation of such a kind of samples by arc melting leads to a non-negligible weight loss (i.e.: as it generally happens for melting and annealing high-melting compounds containing one or more volatile metals), a new synthesis method has also been attempted. also. Even being more laborious technique (since it is a two-step procedure), and requiring sealed containers, it has been successfully employed to prepare a large and single-phase sample (total weight of about 9g) and utilised in the present work. As a first step, the equiatomic binary alloy Mn<sub>2</sub>SbBi has been prepared by induction melting of the elements in outgassed Ta crucibles, closed by welding under pure Ar, by heating up to 1250-1300 °C; manganese antimonide forms by nearly congruent melting, besides, its formation temperature (840 °C) is relatively low: much lower than that of the melting point of Mn metal (1246 °C) [5]. In the second step, Yb and Mn<sub>2</sub>SbBi have been mixed in the stoichiometric amounts, sealed again under Ar into an outgassed Ta crucible, and reacted by induction heating up to about 1300 °C. The crucibles were then sealed under vacuum in quartz tubes and annealed in a resistance furnace at 800 °C for 7 days; after annealing, they were air cooled. Preparation of Mn<sub>2</sub>SbBi inside a Ta crucible has proved not to give rise to pollution contamination due to reaction towards the container material, as well as no Ta contamination in the final YbMn<sub>2</sub>SbBi was noticed.

X-ray powder patterns were obtained either by a Guinier-Stoe camera (Cu Ka radiation, pure Si as an internal standard: *a* = 0.54308 nm), or on a diffractometer DRON-3 (Cu K $\alpha$  radiation, 2 $\Theta$  = 20–70°, 0.05° step, 6 s/step). The Guinier patterns were indexed with the help of the Lazy-Pulverix program [6], the lattice parameters were calculated by least squares method.

The neutron diffraction investigation was carried out from 150 K to 2 K in zero applied field at the Institute Laue-Langevin, Grenoble, France (on the powder D1B diffractometer, wavelength  $\lambda = 0.252$  nm) [7]. The diffraction patterns were indexed, and the calculations performed, by using the Fullprof-program [8].

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#### Table 1

Interatomic distances  $D \pm 5 \times 10^{-4}$  nm and then ratio to sum of the atomic radii [9] of corresponding atom  $\Delta = D/(R_{atom1} + R_{atom2})$  for atoms in La<sub>2</sub>O<sub>2</sub>S-type compounds at 300 K: YbMn<sub>2</sub>Sb<sub>2</sub> (*a* = 0.4528 nm, *c* = 0.7448 nm, *Z*<sub>Mn</sub> = 0.629, *Z*<sub>Sb</sub> = 0.251) and YbMn<sub>2</sub>SbBi (*a* = 0.45493 nm, *c* = 0.7475 nm, *Z*<sub>Mn</sub> = 0.615, *Z*<sub>X</sub> = 0.250).

Atoms	<i>D</i> (nm)	Δ
(a) YbMn <sub>2</sub> Sb <sub>2</sub>		
Yb-6Sb	0.3214	1.01
-6Mn	0.3804 <sup>b</sup>	1.24
–6Yb	0.4528	1.30
Mn–3Sb	0.2763	0.99
–1Sb	0.2815	1.01
–3Mn	0.3244 <sup>c</sup>	1.22
–3Yb	0.3804 <sup>b</sup>	1.24
-6Mn	0.4528 <sup>d</sup>	1.70
Sb-3Mn	0.2763	0.99
–1Mn	0.2815	1.01
–3Yb	0.3214	1.01
(b) YbMn <sub>2</sub> SbBi		
Yb-6X <sup>a</sup>	0.3223	1.00
-6Mn	0.3896 <sup>b</sup>	1.27
–6Yb	0.4549	1.31
Mn-1X	0.2728	0.96
-3X	0.2814	1.00
–3Mn	0.3139 <sup>c</sup>	1.18
–3Yb	0.3896 <sup>b</sup>	1.27
-6Mn	0.4549 <sup>d</sup>	1.70
X–1Mn	0.2728	0.96
-3Mn	0.2814	1.00
–3Yb	0.3223	1.00

<sup>a</sup>  $X = Sb_{0.5}Bi_{0.5}$ .

<sup>b</sup> Yb-Mn<sub>1.2</sub> distance.

<sup>c</sup> Mn<sub>1</sub>-Mn<sub>2</sub> distance.

 $^{d}$  Mn<sub>1</sub>-Mn<sub>1</sub> (Mn<sub>2</sub>-Mn<sub>2</sub>) distance.

### 3. Results and discussion

The substitution of Sb for Bi in YbMn<sub>2</sub>Sb<sub>2</sub> compound leads to changes in cell parameters and interatomic Yb–Mn<sub>1,2</sub>, Mn<sub>1</sub>–Mn<sub>1</sub> (Mn<sub>2</sub>–Mn<sub>2</sub>) and Mn<sub>1</sub>–Mn<sub>2</sub> distances: a = 0.4528 nm, c = 0.7448 nm,  $Z_{Mn} = 0.629$ ,  $Z_{Sb} = 0.251$  in YbMn<sub>2</sub>Sb<sub>2</sub> and a = 0.45493 nm, c = 0.7475 nm,  $Z_{Mn} = 0.615$ ,  $Z_X = 0.250$  in YbMn<sub>2</sub>SbBi (Table 1). The ratio of shortest interatomic distances to sum of the atomic radii of Yb, Mn, Sb and Bi atom  $\Delta = D/(R_{atom1} + R_{atom2})$  corresponds to the metallic nature of bond and trivalet state of Yb in the YbMn<sub>2</sub>Sb<sub>2</sub> and YbMn<sub>2</sub>SbBi (R<sub>Yb</sub><sup>III</sup> = 0.1740 nm, R<sub>Mn</sub> = 0.1334 nm, R<sub>Sb</sub> = 0.1451 nm, R<sub>Bi</sub> = 0.1537 nm [9]), The Mn<sub>1</sub>–Mn<sub>2</sub> distances decreases from YbMn<sub>2</sub>Sb<sub>2</sub> to YbMn<sub>2</sub>SbBi with distortion of Mn coordination polyhedron.

In general, the symmetry operation of  $\mathbf{D_{3d}}$  ( $\bar{3}2/m$ ) point group includes  $\mathbf{S_6}$  ( $\bar{3}$ ),  $\mathbf{D_3}$  ( $\mathbf{32}$ ),  $\mathbf{C_{3v}}$  ( $\mathbf{3m}$ ) subgroups of index 2,  $\mathbf{D_{1d}}$  (2/m) subgroup of index 3,  $\mathbf{C_3}$  ( $\mathbf{3}$ ) subgroup of index 4 and  $\mathbf{C_{1v}}$  ( $\mathbf{m}$ ),  $\mathbf{C_i}$ ( $\mathbf{1'}$ ) and  $\mathbf{C_2}$  ( $\mathbf{2}$ ) subgroups of index 6 [10]. The possible magnetic point groups for the  $\mathbf{D_{3d}}$  ( $\bar{3}2/m$ ) point group are above-mentioned "colourless" point groups and "black-white" (Shubnikov) { $\mathbf{S_6}$ ,  $2_x \times \mathbf{S_6} \times 1'$ }, { $\mathbf{D_3}$ ,  $m_x \times \mathbf{D_3} \times 1'$ } and { $\mathbf{C_{3v}}$ ,  $2_x \times \mathbf{C_{3v}} \times 1'$ }. The magnetic structure of the La<sub>2</sub>O<sub>2</sub>S-type compound was determined in terms of the symmetry of above-mentioned magnetic point groups.

The evolution of the magnetic order upon cooling shows the development of commensurate magnetic reflections with wave vector  $\mathbf{K}_0 = [0, 0, 0]$  in the YbMn<sub>2</sub>SbBi neutron diffraction patterns (Fig. 1). Thermal variation of magnetic reflections shows the magnetic ordering below ~138 K (Fig. 2a). Analysis of the diffraction data shows that between 138 and 112 K, the magnetic structure of YbMn<sub>2</sub>SbBi consists of *ab*-plane antiferromagnetically coupled magnetic moments of the manganese atoms that normal to *a* axis (**D**<sub>1d</sub> magnetic point group, **P2/m** magnetic space group, **AF**<sup>K0</sup> (**D**<sub>1d</sub>) antiferromagnetic component) (Fig. 3a). Below







Fig. 2. Thermal evaluation of some reflections (a) and Yb and Mn magnetic components (b) of the  $YbMn_2SbBi$ .

#### Table 2

Crystallographic and magnetic parameters of La<sub>2</sub>O<sub>2</sub>S-type YbMn<sub>2</sub>SbBi compound at different temperatures *T*: the temperatures of magnetic ordering  $T_{N,C}$ , cell parameters *a*, *c*, atomic position parameters  $Z_{Mn}$  and  $Z_X$ , ferromagnetic and antiferromagnetic component  $M_j^F$  and  $M_j^{AF}$  of the *j* atom (Yb, Mn1<sup>a</sup> and Mn2),  $\varphi_j^{AF}$  and  $\varphi_j^F$  are the angles of corresponding magnetic component with *a* axis. The magnetic moments lie in the *ab* plane: the  $\theta$  angle of all magnetic components with *c* axis is 90°. Reliability factors  $R_F$  (crystal structure) and  $R_F^m$  (magnetic structure) are given in percent (%).

T <sub>N,C</sub>	Magnetic ordering	<i>T</i> (K)	Unit cell data	R <sub>F</sub>	Atom	$M_{j}^{F}(\mu_{B})$	$\varphi_{j}{}^{\mathrm{F}}\left(^{\circ} ight)$	$M_j^{AF}(\mu_B)$	$arphi_j{}^{ m AF}$ ( $^\circ$ )	$R_{\rm F}{}^{\rm m}$
	Paramagnet	150	a = 0.45493(6)  nm c = 0.7475(2)  nm $Z_{Mn} = 0.615(2)$ $Z_X = 0.250(2)$	5.2						
$T_{\rm N}^{\rm ND}$ = 138(3) K	Antiferromagnet $AF^{K0}(D_{1d})$	118	a = 0.45493(6)  nm c = 0.7464(2)  nm $Z_{Mn} = 0.615(2)$ $Z_X = 0.249(2)$	4.5	Yb Mn1 Mn2			0 2.2(1) 2.2(1)	270 90	7.3
$T_{\rm C}^{\rm ND} = 112(3)  {\rm K}$	Ferrimagnet $\mathbf{AF}^{K0} (\mathbf{D_{1d}}) + \mathbf{F}_{a}^{K0} (\mathbf{C}_{2})$	2	a = 0.45490(5)  nm c = 0.7448(2)  nm $Z_{Mn} = 0.597(4)$ $Z_X = 0.236(2)$	4.5	Yb Mn1 Mn2	3.6(3) 1.4(2) 1.4(2)	0 0 0	3.2(2) 3.2(2)	270 90	6.9

<sup>a</sup> Mn1 occupy  $(1/3, 2/3, X_{Mn})$  site and Mn2 occupy  $(2/3, 1/3, -X_{Mn})$  site in the La<sub>2</sub>O<sub>2</sub>S-type unit cell.

112(3) K, the ferromagnetic components of Yb and Mn begin to develop, and the ferromagnetic transition occurs with resulting non-collinear arrangements of magnetic moments in the *ab* plane (Fig. 3b). The magnetic structure of YbMn<sub>2</sub>SbBi becames the sum antiferromagnetic component **AF**<sup>K0</sup> (**D**<sub>1d</sub>) and ferromagnetic one with **C**<sub>2</sub> magnetic point group (**F**<sub>a</sub><sup>K0</sup> (**C**<sub>2</sub>)). The crystallographic data and magnetic parameters of YbMn<sub>2</sub>SbBi are given in Table 2. The maximum of the I<sub>(002)</sub> intensity near ~70 K (Fig. 2a) occurs when ferromagnetic component of YbMn<sub>2</sub>SbBi became bigger than antiferromagnetic one (Fig. 2b). The magnitude of Yb and Mn magnetic moments at 2 K (M<sub>Yb</sub> = 3.6(2) µ<sub>B</sub>, M<sub>Mn</sub> = 3.5(2) µ<sub>B</sub>) correspond to



Fig. 3. Magnetic structure of the YbMn\_2SbBi compound between 138 and 112 K (a) and below 112 K (b).

the trivalent state of the Yb ions  $(M_{Yb3+} = 4.0 \mu_B)$  and tetravalent state of the Mn ions  $(M_{Mn4+} = 4.0 \mu_B)$  [4].

In general, the evaluation of magnetic ordering of YbMn<sub>2</sub>SbBi is following: paramagnet (symmetry  $D_{3d}$  point group,  $P\bar{3}ml$  space group)  $\rightarrow$  Antiferromagnet (symmetry  $D_{1d}$ , P2/m)  $\rightarrow$  Ferrimagnet (symmetry  $C_2$ , P2). The YbMn<sub>2</sub>Sb<sub>2</sub> shows evaluation of magnetic ordering: paramagnet (symmetry  $D_{3d}$  point group,  $P\bar{3}ml$  space group)  $\rightarrow$  Antiferromagnet (symmetry  $D_{3d}$  point group,  $P\bar{3}ml$  space group)  $\rightarrow$  Antiferromagnet (symmetry  $D_{1d}$  ore  $C_i$  magnetic points group) [3]. So, the presence of Bi in the p-element mixed sublattice (Sb and Bi) facilitates the ferromagnetic ordering of La<sub>2</sub>O<sub>2</sub>S-type compound.

### 4. Conclusion

The applied magnetic fields may play a role in shifting the magnetic transition temperature as it happens in the  $YbMn_2Sb_2$  compound. In the known homologous compound  $EuMn_2Sb_2$  [11], though the magnetic moments of the Mn-atoms sublattice might order antiferromagnetically, the overall magnetic structure could however be much more complicated, due to the possible ordering of the Eu-sublattice.

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